The Stereoselective Total Synthesis of (6S)-5,6-Dihydro-6-[(2R)-2-hydroxy-6-phenylhexyl]-2H-pyran-2-one via Prins Cyclization

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The stereoselective total synthesis of an antiproliferative and antifungal α -pyrone natural product (6S)-5,6-dihydro-6-[(2R)-2-hydroxy-6-phenylhexyl]-2H-pyran-2-one is described. The key steps involved are the *Prins* cyclization, *Mitsunobu* reaction, and ring-closing metathesis reaction.

Introduction. – Antiproliferative and antifungal α -pyrone natural product (6*S*)-5,6-dihydro-6-[(2*R*)-2-hydroxy-6-phenylhexyl]-2*H*-pyran-2-one (1) was isolated by *Hostettmann* and co-workers in 2001 from *Ravensara crassifolia* [1]. The α -pyrone moiety is one of the most commonly encountered structural motif among the various natural-product skeletons. Many of these natural products exhibit varied pharmacological properties such as antimicrobial, antifungal, and cytotoxicity against human tumor cells [2]. Some of these compounds such as fostriecin [3] and goniothalamin [3] are known to display anticancer properties, and (–)-pyronetin [4] is known to exhibit immuno-suppressive activity. Passifloricin-A [5] and strictifolione [6] are found to exhibit antifungal activity. (6*S*)-5,6-Dihydro-6-[(2*R*)-2-hydroxy-6-phenylhexyl]-2*H*-pyran-2-one (1) is also one such natural product which is known to exhibit antifungal activity against phytopathogenic fungus *Cladosporium cucumarinum* (*Fig.*) [1]. In a recent investigation, this compound was found to display antiproliferative activity against three human cancer cell lines: leukemia (THP-1 and U-937) and melanoma (A-375) [7].

Figure. Structure of (6S)-5,6-Dihydro-6-[(2R)-2-hydroxy-6-phenyl-hexyl]-2H-pyran-2-one (1)

The *Prins* cyclization is a powerful synthetic tool for the construction of tetrahydropyran scaffolds and has been utilized in the synthesis of several natural products [8]. Our group has made a significant effort to explore the utility of *Prins* cyclization in the synthesis of various polyketide intermediates and applied it to the total synthesis of some natural products [9]. The structural uniqueness of compound 1,

coupled with interesting biological activity and our interest on the 6-substituted α,β -unsaturated- δ -lactones [10], prompted us to explore the total synthesis via the Prins cyclization. In our retrosynthetic analysis, we envisaged that the target molecule 1 could be achieved from the diene ester 2 by a ring-closing metathesis approach. The ester 2 can be easily prepared from 2,4,6-trisubstituted tetrahydropyran 3, which could in turn be obtained via the Prins cyclization of homoallylic alcohol 4 with aldehyde 5 (Scheme~1).

Scheme 1. Retrosynthetic Analysis of Compound (1)

$$1 \Longrightarrow \bigvee_{\text{Ph}(\text{CH}_2)_4} \bigvee_{\text{O}} \bigvee_{\text{O}} \bigvee_{\text{Ph}} \bigvee_{\text{CHO}} \bigvee_{\text{Ph}} \bigvee_{\text{CHO}} \bigvee_{\text{CHO}} \bigvee_{\text{Ph}} \bigvee_{\text{CHO}} \bigvee_{\text{CHO}} \bigvee_{\text{Ph}} \bigvee_{\text{CHO}} \bigvee_{$$

Results and Discussion. – The synthesis of the natural product **1** started with chiral homoallyl alcohol **4** (*Scheme* 2). The precursor **4** was prepared in two steps by Cu^{I} -mediated regioselective opening [11] of (–)-(R)-2-[(benzyloxy)methyl]oxirane (**6**) with vinyl magnesium bromide, followed by subsequent reductive cleavage of benzyloxy alcohol **7** with Li in liquid NH_3 . The *Prins* cyclization of **4** with 5-phenylpentanal (**5**) in the presence of TFA (10 equiv.), followed by hydrolysis of the resulting trifluoroacetate, gave the trisubstituted pyran **8** in 50% yield [12]. The

a) CH₂=CHMgBr, CuCN, -78° to -40° , 4 h; 92%. b) Li, liquid NH₃, THF, 20 min; 75%. c) 5-Phenylpentanal (5), CF₃COOH (TFA), CH₂Cl₂, then K₂CO₃, MeOH, r.t., 3 h; 50%. d) Et₃N, TsCl, CH₂Cl₂, 0° to r.t., 3 h; 96%. e) (t-Bu)Me₂SiCl (TBSCl), 4-(dimethylamino)pyridine (DMAP), 1H-imidazole, CH₂Cl₂, 0° to r.t., 3 h; 95%. f) NaI, acetone, reflux, 24 h; 95%. g) Zn, EtOH, reflux, 1 h; 98%.

stereochemistry of step c leading to $\mathbf{8}$ was assumed to be in accordance with previous results [9]. However, it was later established after elaborating compound $\mathbf{8}$ to the target molecule $\mathbf{1}$, which in all respects was identical with that described in literature [1].

The chemoselective tosylation of primary alcohol **8** with 1.1 equiv. of TsCl in the presence of Et₃N in CH₂Cl₂ gave the corresponding tosyl derivative **9** in 96% yield [13]. TBS Protection of the secondary alcohol **9** with TBSCl, DMAP, and 1*H*-imidazole afforded the TBS ether **10** in 95% yield. Treatment of **10** with NaI in refluxing acetone gave the corresponding iodo compound **3** in 95% yield, which, on exposure to activated Zn in refluxing EtOH, furnished key intermediate **11**.

Secondary alcohol **11** was protected as its methoxymethyl (MOM) ether **12** using MOM-Cl and EtN(i-Pr)₂ in CH₂Cl₂ (*Scheme 3*). Deprotection of TBS ether **12** with Bu₄NF in THF gave the alcohol **13** in 90% yield, which was subjected to standard *Mitsunobu* reaction conditions [14] using diethyl azodicarboxylate (DEAD), PPh₃, and 4-nitrobenzoic acid, followed by hydrolysis of the resulting ester with K₂CO₃ in MeOH, to give the homoallylic alcohol **14** with inversion of configuration. Homoallyl alcohol **14** was then subjected to esterification with acryloyl chloride to obtain the diene ester **2**. The obtained ester **2** was used for the ring-closing metathesis (RCM) with *Grubbs*' first-generation catalyst (5 mol-%) to provide the desired lactone **15** [15]. Finally, deprotection of MOM ether **15** was achieved with 6N HCl/THF/H₂O 1:2:1 to afford the target natural product **1**. The spectral and physical data of all the compounds were in good agreement with literature values [1].

a) Methoxymethyl chloride (MOM-Cl), DIPEA, CH_2Cl_2 , 4 h, 0°; 93%. b) Bu_4NF , THF, 0° to r.t., 2 h; 90%. c) 4-Nitrobenzoic acid, diethyl azodicarboxylate (DEAD), PPh₃ then K_2CO_3 , MeOH, r.t., 0.5 h; 86%. d) Acryloyl chloride, $EtN(i-Pr)_2$, CH_2Cl_2 , 0° to r.t.; 83%. e) Grubbs' I generation catalyst [(PCy₃)₂Cl₂Ru=CHPh] (5 mol-%), dry CH_2Cl_2 , 55°, 12 h; 90%. f) 6N HCl/THF/H₂O 1:2:1, r.t., 24 h; 92%.

In conclusion, we have established the versatility of the *Prins* cyclization and a subsequent reductive cleavage in natural-product synthesis by achieving the stereo-

selective synthesis of the antiproliferative and antifungal natural product (6S)-5,6-dihydro-6-[(2R)-2-hydroxy-6-phenylhexyl]-2H-pyran-2-one (1). To the best of our knowledge, this is the first report on the total synthesis of 1 by means of the *Prins* cyclization.

Experimental Part

General. Solvents were dried over standard drying agents and freshly distilled prior to use. The reagents were purchased from the *Aldrich* and *Acros*, and were used without further purification unless otherwise stated. All moisture-sensitive reactions were carried out under N₂. Org. solns. were dried over anh. Na₂SO₄ and concentrated *in vacuo* below 40°. Column chromatography (CC): silica gel (SiO₂; 60–120 mesh, *Acme*). Optical rotations: *Horiba* high-sensitive polarimeter *SEPA-300* at 25°. IR Spectra: *Perkin-Elmer IR-683* spectrophotometer with NaCl optics. ¹H- (300 MHz) and ¹³C-NMR (75 MHz) spectra: *Bruker Avance 300* instrument with TMS as internal standard in CDCl₃; *J* values in Hz. MS: *Agilent Technologies 1100 Series* (Agilent Chemistation Software).

(1R)-1,5-Anhydro-2,4-dideoxy-1-(4-phenylbutyl)-L-threo-hexitol (8). Trifluoroacetic acid (TFA; 23.4 ml, 306 mmol) was added to a soln. of homoallylic alcohol 4 (1.04 g, 10.18 mmol) and 5-phenylpentanal (5 g, 30.6 mmol) in dry CH₂Cl₂ (30 ml) under N₂. The progress of the reaction was followed by TLC. Upon completion, the reaction was quenched with aq. NaHCO₃ soln., and the pH was adjusted to >7 by addition of Et₃N. The aq. phase was extracted with CH₂Cl₂ (3 × 20 ml), and the combined org. layers were dried (MgSO₄) and concentrated under reduced pressure. The resulting crude trifluoroacetate was then dissolved in MeOH (20 ml) and treated with K₂CO₃ (770 mg, 7.2 mmol) overnight. After complete conversion, MeOH was removed under reduced pressure, and then the reaction was quenched with H₂O (20 ml), and the mixture was extracted with CH₂Cl₂ (3 × 25 ml). The combined org. layers were dried (MgSO₄) and concentrated *in vacuo*. The product was then purified by flash CC to afford 8 (310 mg, 50% yield over two steps). White solid. [α]²⁰₀ = -1.1 (c = 1.4, CHCl₃). IR (KBr): 3310, 3083, 3025, 2929, 2855, 1941, 1702, 1605, 1472, 1369, 1260, 1159, 1077, 1018, 934, 828, 694, 635. ¹H-NMR: 7.29 – 7.05 (m, 5 H); 3.85 – 3.65 (m, 1 H); 3.61 – 3.21 (m, 4 H); 2.60 (t, t = 7.5, 2 H); 2.09 – 1.72 (t = 4, 4); 1.71 – 1.31 (t = 5, 5, 7; 31.3; 25.0. LC/MS: 287 ([t = 1, 4)]

(1R)-1,5-Anhydro-2,4-dideoxy-6-O-[(4-methylphenyl)sulfonyl]-1-(4-phenylbutyl)-L-threo-hexitol (9). To a soln. of **8** (300 mg, 1.16 mmol) in dry CH₂Cl₂ (10.0 ml), Et₃N (0.32 ml, 2.32 mmol) was added at 0°. Then, TsCl (265 mg, 1.39 mmol) was added over 2 h. The resulting mixture was allowed to stir at r.t. for 3 h. Then, the mixture was treated with 1M aq. HCl (4 ml) and extracted with CH₂Cl₂ (3 × 5 ml). The org. layer was washed with sat. NaHCO₃ (5 ml) and H₂O (5 ml). The combined org. phases were dried (Na₂SO₄) and concentrated under reduced pressure. Flash chromatography (FC) of the crude product afforded **9**. [α]₀²⁰ = +13.2 (c = 0.55, CHCl₃). ¹H-NMR: 7.76 (d, J = 8.0, 2 H); 7.34 – 7.07 (m, 7 H); 4.01 – 3.90 (m, 1 H); 3.80 – 3.63 (m, 1 H); 3.58 – 3.42 (m, 1 H); 3.30 – 3.13 (m, 1 H); 2.63 – 2.53 (m, 2 H); 2.45 (n, 3 H); 1.96 – 1.69 (n, 3 H); 1.67 – 1.48 (n, 2 H); 1.47 – 1.20 (n, 4 H); 1.19 – 0.86 (n, 2 H). ¹³C-NMR: 144.7; 142.4; 132.8; 129.7; 128.3; 128.2; 127.8; 125.5; 75.6; 72.7; 71.9; 67.5; 40.6; 36.8; 35.7; 35.5; 31.3; 25.0; 21.5. LC/MS: 441 ([M + Na]⁺).

(1R)-1,5-Anhydro-3-O-[(tert-butyl)(dimethyl)silyl]-2,4-dideoxy-6-O-[(4-methylphenyl)sulfonyl]-1-(4-phenylbutyl)-L-threo-hexitol (10). To a soln. of $\bf 9$ (300 mg, 0.71 mmol) in dry CH₂Cl₂ (5 ml) and 1*H*-imidazole (98 mg, 1.43 mmol) at 0° under N₂ was added (*t*-Bu)Me₂SiCl (TBSCl; 215 mg, 1.43 mmol), and the resulting mixture was allowed to stir for 3 h at r.t. Then, the mixture was diluted with H₂O (3 ml) and extracted with CH₂Cl₂ (2 × 5 ml). The combined org. layers were washed with brine (1 × 5 ml), dried (Na₂SO₄), and concentrated under vacuum to afford the crude product. CC of the crude product gave $\bf 10$. Colorless liquid. [α]²⁰_D = +2.3 (c = 1.15, CHCl₃). ¹H-NMR: 7.97 – 7.94 (d, d = 8.3, 2 H); 7.50 – 7.36 (d m, 5 H); 7.33 – 7.26 (d m, 2 H); 4.15 – 4.08 (d m, 2 H); 3.91 – 3.80 (d m, 1 H); 3.72 – 3.62 (d m, 1 H); 3.41 – 3.31 (d m, 1 H); 2.76 (d m, 2 H); 2.62 (d m, 3 H); 1.95 – 1.83 (d m, 2 H); 1.82 – 1.70 (d m, 2 H); 1.69 – 1.39 (d m, 4 H); 1.37 – 1.22 (d m, 2 H); 1.03 (d m, 3.8; 35.6; 31.4; 25.7; 25.0; 21.6; 17.9; – 4.5. LC/MS: 555 ([d H Na]⁺).

(tert-Butyl){[(2R,4S,6R)-2-(iodomethyl)-6-(4-phenylbutyl)tetrahydro-2H-pyran-4-yl]oxy}dimethyl-silane (3). NaI (493 mg, 3.28 mmol) was added to a soln. of **10** (350 mg, 0.657 mmol) in 10 ml of acetone and heated to reflux for 24 h. Acetone was removed under reduced pressure, the residue was diluted with H₂O and AcOEt, and the org. layer was separated, dried (Na₂SO₄), and concentrated *in vacuo*. The resulting crude product was purified by CC to afford **3**. Colorless liquid. [α] $_0^{20}$ = -11.4 (c = 1.1, CHCl₃). IR (KBr): 3449, 3025, 2930, 2855, 1636, 1495, 1461, 1254, 1130, 1073, 838, 774, 746, 697. ¹H-NMR: 7.45 – 7.23 (m, 5 H); 3.95 – 3.82 (m, 1 H); 3.51 – 3.36 (m, 1 H); 3.35 – 3.24 (m, 2 H); 2.80 – 2.73 (m, 2 H); 2.21 – 2.11 (m, 2 H); 1.89 – 1.48 (m, 6 H); 1.46 – 1.23 (m, 2 H); 1.03 (m, 9 H); 0.20 (m, 6 H). ¹³C-NMR: 142.7; 128.3; 128.3; 125.5; 75.8; 75.2; 68.4; 41.3; 41.2; 35.9; 35.7; 31.4; 25.8; 25.3; 18.0; 9.0; – 4.5. LC/MS: 489 (m) m) m)

(5R,7R)-7-[[(tert-Butyl)(dimethyl)silyl]oxy]-1-phenyldec-9-en-5-ol (**11**). To **3** (300 mg, 0.613 mmol) in EtOH (10 ml), commercial Zn dust (66.3 mg, 9.20 mmol) was added, and the resulting mixture was refluxed for 1 h and then cooled to 25°. Addition of solid NH₄Cl and Et₂O (15 ml), followed by stirring for 5 min, gave a gray suspension which was filtered through *Celite*, and the filtrate was concentrated *in vacuo*. The resulting crude product was purified by FC to give **11**. Colorless liquid. [α]_D²⁰ = -10.6 (c = 0.35, CHCl₃). IR (KBr): 3456, 3073, 2931, 2856, 1640, 1462, 1365, 1253, 1072, 1000, 912, 834, 775, 742, 697.

¹H-NMR: 7.28-7.07 (m, 5 H); 5.78-5.62 (m, 1 H); 5.09-4.99 (m, 2 H); 4.05-3.96 (m, 1 H); 3.93-3.82 (m, 1 H); 2.89 (br. s, 1 H); 2.60 (t, t = 7.5, 2 H); 2.36-2.28 (m, 2 H); 1.68-1.23 (m, 8 H); 0.90 (s, 9 H); 0.09 (s, 3 H); 0.08 (s, 3 H). ¹³C-NMR: 142.6; 134.6; 128.3; 128.1; 125.8; 117.3; 71.2; 68.0; 41.1; 40.9; 37.7; 35.9; 31.5; 25.7; 25.2; 17.8; -4.5; -4.8. LC/MS: 363 ([m + H] $^+$).

(4S,6R)-6-(Methoxymethoxy)-10-phenyldec-1-en-4-ol (14). To a stirred mixture of 13 (200 mg, 0.492 mmol), Ph₃P (387 g, 1.47 mmol), and 4-nitrobenzoic acid (1.3 g, 0.639 mmol) in dry THF (5 ml) at 0° was added diethyl azodicarboxylate (DEAD; 0.18 ml, 1.47 mmol) via a syringe. The mixture was then stirred for another 0.5 h at r.t., then diluted with H₂O (5 ml), and extracted with AcOEt (2 × 15 ml). After removal of the solvent, the crude residue was dissolved in MeOH (5 ml) and treated with K₂CO₃ (66 mg, 0.96 mmol). After stirring the mixture for 4 h at r.t., it was diluted with H₂O (5 ml) and extracted with CH₂Cl₂ (3 × 5 ml). Removal of solvent under reduced pressure, followed by FC, afforded 14. Colourless liquid. [a] $_{20}^{20}$ = -25.4 (c = 1.15, CHCl₃). 1 H-NMR: 7.27 -7.07 (m, 5 H); 5.86 -5.74 (m, 1 H); 5.11 -5.03 (m, 2 H); 4.59 (s, 2 H); 3.87 -3.80 (m, 1 H); 3.78 -3.72 (m, 1 H); 3.34 (s, 3 H); 2.61 -2.57 (m, 2 H); 2.21 -2.15 (m, 2 H); 1.69 -1.45 (m, 7 H); 1.42 -1.29 (m, 2 H). 13 C-NMR: 142.1; 134.8; 128.7; 125.8; 117.5; 95.3; 71.1; 67.17; 55.5; 41.9; 40.8; 36.6; 34.1; 31.4; 30.6; 24.3. LC/MS: 292 (m⁺).

(1S,3R)-3-(Methoxymethoxy)-7-phenyl-1-(prop-2-en-1-yl)hept-1-yl Prop-2-enoate (2). To a stirred soln. of **14** (200 mg, 0.69 mmol) in dry CH₂Cl₂ (5 ml) were added acryloyl chloride (0.1 ml, 1.37 mmol) and EtN(i-Pr)₂ (0.4 ml, 2.748 mmol) at 0°. The resulting mixture was allowed to stir at r.t. for 3 h. The mixture was then diluted with H₂O (5 ml) and extracted with CH₂Cl₂ (3 × 5 ml). The combined org. layers were dried (Na₂SO₄), and the solvent was removed under vacuum to give the crude residue, which was purified by SiO₂ CC to afford pure **2**. [α]_D²⁰ = +25.1 (c = 1.0, CHCl₃). IR (KBr): 3506, 3432, 1926, 1733, 1450, 1098, 985, 746. ¹H-NMR (200 MHz, *Varian gemini FT-200*): 7.21 (t, J = 7.5, 2 H); 7.11 (t, J = 5.2, 3 H); 6.36 (d, J = 7.3, 1 H); 6.12 – 6.0 (m, 1 H); 5.82 – 5.65 (m, 2 H); 5.15 – 5.01 (m, 3 H); 4.63 – 4.50 (m, 2 H); 3.63 – 3.49 (m, 1 H); 3.39 – 3.28 (m, 3 H); 2.59 (t, J = 7.5, 2 H); 2.45 – 2.26 (m, 2 H); 2.0 – 1.23 (m, 8 H). FAB-MS: 285 ([M + OCH₂ – OMe]⁺).

(6S)-5,6-dihydro-6-[(2R)-2-(methoxymethoxy)-6-phenylhexyl]-2H-pyran-2-one (15). To a stirred soln. of bis(tricyclohexylphosphine)benzylideneruthenium(IV) dichloride (*Grubbs*' catalyst; 22 mg, 5 mol-%) in CH₂Cl₂ (50 ml) at 55° was added a soln. of 14 (190 mg, 0.54 mmol) in CH₂Cl₂ (10 ml). The resulting mixture was heated for 12 h. After completion of the reaction, the contents were cooled, and the solvent was removed under reduced pressure to yield a crude product, which was purified by SiO₂ CC to afford pure 15. [a] $_{20}^{20}$ = -38.6 (c = 0.65, CHCl $_{3}$). IR (KBr): 2933, 1724, 1386, 1299, 1034, 750. 1 H-NMR: 7.22 (t, J = 8.3, 2 H); 7.11 (t, J = 7.1, 3 H); 6.87 – 6.79 (m, 1 H); 5.99 (d, J = 9.4, 1 H); 4.63 – 4.49 (m, 3 H); 3.75 – 3.66 (m, 1 H); 3.30 (s, 3 H); 2.60 (t, J = 7.5, 2 H); 2.41 – 2.25 (m, 2 H); 2.15 – 2.02 (m, 1 H); 1.70 (m, 5 H). LC/MS: 318 (M⁺).

(6S)-5,6-Dihydro-6-[(2R)-2-hydroxy-6-phenylhexyl]-2H-pyran-2-one (1). Compound 15 (150 mg, 0.414 mmol) was stirred in 4 ml of 6n HCl/THF/H₂O 1:2:1 for 24 h. The reaction was quenched with sat. NaHCO₃ (5 ml) and H₂O (5 ml), and the mixture was extracted with AcOEt. The combined org. layers

were washed with brine $(1 \times 5 \text{ ml})$, dried (Na_2SO_4) , and concentrated *in vacuo*. The resulting crude product was purified by CC to afford **1**. Pale-yellow solid. M.p. $33-35^\circ$. $[a]_D^{20}=-64.9$ $(c=0.65,\text{CHCl}_3)$. IR (KBr): 3457, 2938, 2848, 1695, 1487, 1392, 1254. ¹H-NMR: 7.33-7.12 (m, 5 H); 6.94-6.84 (m, 1 H); 6.02 (dd, J=1.4, 9.4, 1 H); 4.82-4.65 (m, 1 H); 4.07-3.93 (m, 1 H); 2.62 (t, J=7.2, 2 H); 2.40-2.30 (m, 2 H); 1.97-1.81 (m, 1 H); 1.70-1.58 (m, 3 H); 1.52-1.38 (m, 4 H). ¹³C-NMR: 164.4; 145.4; 142.3; 128.2; 128.1; 125.5; 121.2; 74.9; 66.8; 42.2; 37.7; 35.8; 31.3; 29.8; 25.1. LC/MS: 275 $([M+H]^+)$.

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